

UNIVERSITY OF ESWATINI
MAIN EXAMINATION 2018/2019

TITLE OF PAPER: CHEMICAL APPLICATIONS OF
GROUP THEORY

COURSE NUMBER: CHE321

TIME ALLOWED: THREE (3) HOURS

INSTRUCTIONS: THERE ARE FOUR (4) QUESTIONS IN
THIS PAPER. ANSWER QUESTION
ONE (TOTAL 40 MARKS) AND ANY
TWO OTHER QUESTIONS (EACH
QUESTION IS 30 MARKS)

A PERIODIC TABLE AND OTHER USEFUL DATA HAVE BEEN
PROVIDED WITH THIS EXAMINATION PAPER

PLEASE DO NOT OPEN THIS PAPER UNTIL AUTHORISED TO
DO SO BY THE CHIEF INVIGILATOR.

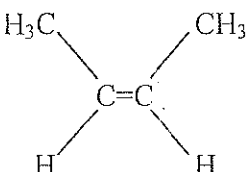
QUESTION ONE (COMPULSORY) [40 Marks]

(a) Draw the shapes of the following species and state the number of electron lone pairs:

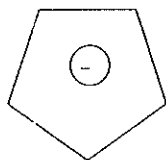
(i) FeF_4 (ii) BH_4^- (iii) NF_4^+ [6]

(b) List all the symmetry elements of the following molecules:

(i) ClF_3

(ii)  (assume CH_3 is spherical)

(iii)



[6]

(c) Classify the following species into their point groups:

(i) Cyclopropane

(ii) *trans*- N_2F_2

(iii) CO_2

[9]

(d) (i) Set up the matrices which will perform the following transformations:

$$(1) \begin{pmatrix} x \\ y \end{pmatrix} \quad \text{to} \quad \begin{pmatrix} -y \\ -x \end{pmatrix}$$

$$(2) \begin{pmatrix} x \\ y \\ z \end{pmatrix} \quad \text{to} \quad \begin{pmatrix} -y \\ x \\ -z \end{pmatrix}$$

[4]

(ii) Which of the following molecules or ions contain

(1) a C_3 axis but no σ_h plane

(2) a C_3 axis and a σ_h plane:

NH_3 ; SO_3 ; PBr_3 ; AlCl_3 ; $[\text{SO}_4]^{2-}$; $[\text{NO}_3]^-$

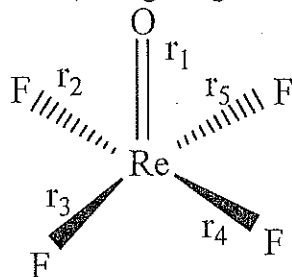
[6]

(e) With the help of group theory methods determine the number of IR and Raman peaks expected for SiF_4 .

[9]

QUESTION TWO [30 Marks]

- (a) (i) By substituting H's with Cl's in CH_4 you obtain CH_3Cl , CH_2Cl_2 , CHCl_3 and CCl_4 . Give the point groups of these four substituted molecules. [8]
- (ii) How many planes of symmetry do the following molecules possess?
 (1) $\text{F}_2\text{C}=\text{O}$ (2) $\text{ClFC}=\text{O}$ [2]
- (b) The structure of tetrafluorooxorhenium(VI), ReOF_4 (C_{4v} symmetry), can be diagrammed as below. Use the accompanying C_{4v} character table to carry out the following tasks. Let the basis set for internal bond displacement coordinates be r_1, r_2, r_3, r_4, r_5 with r_1 being assigned to the $\text{Re}=\text{O}$ bond.



Using internal coordinates, determine the total reducible representation for Re-F ligand stretching modes and decompose it into irreducible representations. [5]

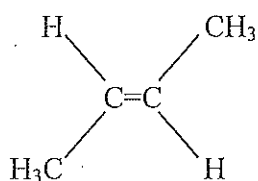
- (i) Determine allowed IR and Raman bands for the molecule. [5]
- (c) With an example distinguish between
 (i) a symmetry element and a symmetry operation. [5]
 (ii) symmetry operations of the class and equivalent symmetry elements. [5]

QUESTION THREE [30 Marks]

- (a) Using group theory methods determine the hybrid orbital schemes on the central atom in $[\text{NbF}_5]$ (square pyramid) and select the most suitable orbital set for bonding. Use Nb-F bonds as a basis. [10]
- (b) For the following octahedral-based compounds, where M is a central atom, A and B are distinct monodentate ligands and (A^A) is a chelating bidentate ligand, name the point group to which each of the following species belong.
 (i) $M(A^A)B_4$ (ii) trans-MA_2B_4 (iii) cis-MA_2B_4 [9]
- (c) (i) Reduce the following representation [4]
- | | | | | | |
|-------|---|--------|--------|--------|-------------|
| T_d | E | $8C_3$ | $3C_2$ | $6S_4$ | $6\sigma_d$ |
| | 4 | 1 | 0 | 0 | 2 |
- (ii) Sketch a qualitative molecular orbital energy level diagram for H_2O molecule using group theory methods. [7]

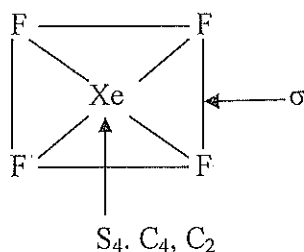
QUESTION FOUR [30 Marks]

- (a) (i) Deduce which symmetry elements are lost on going from
 (1) BF_3 to BClF_2
 (2) BClF_2 to BBrClF
 (3) Which symmetry element (apart from E) is common to all three molecules above? [6]
- (ii) Set up the multiplication table for the operations of the molecule *trans*-but-2-ene. Apply the top operation then the side operation. [4]



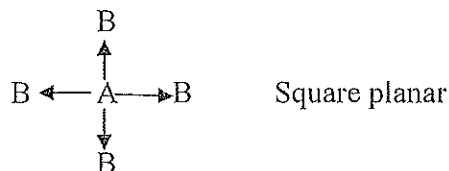
	E	C_2	σ	i
E				
C_2				
σ				
i				

- (b) (i) The diagram below shows the location of the symmetry elements in XeF_4 .



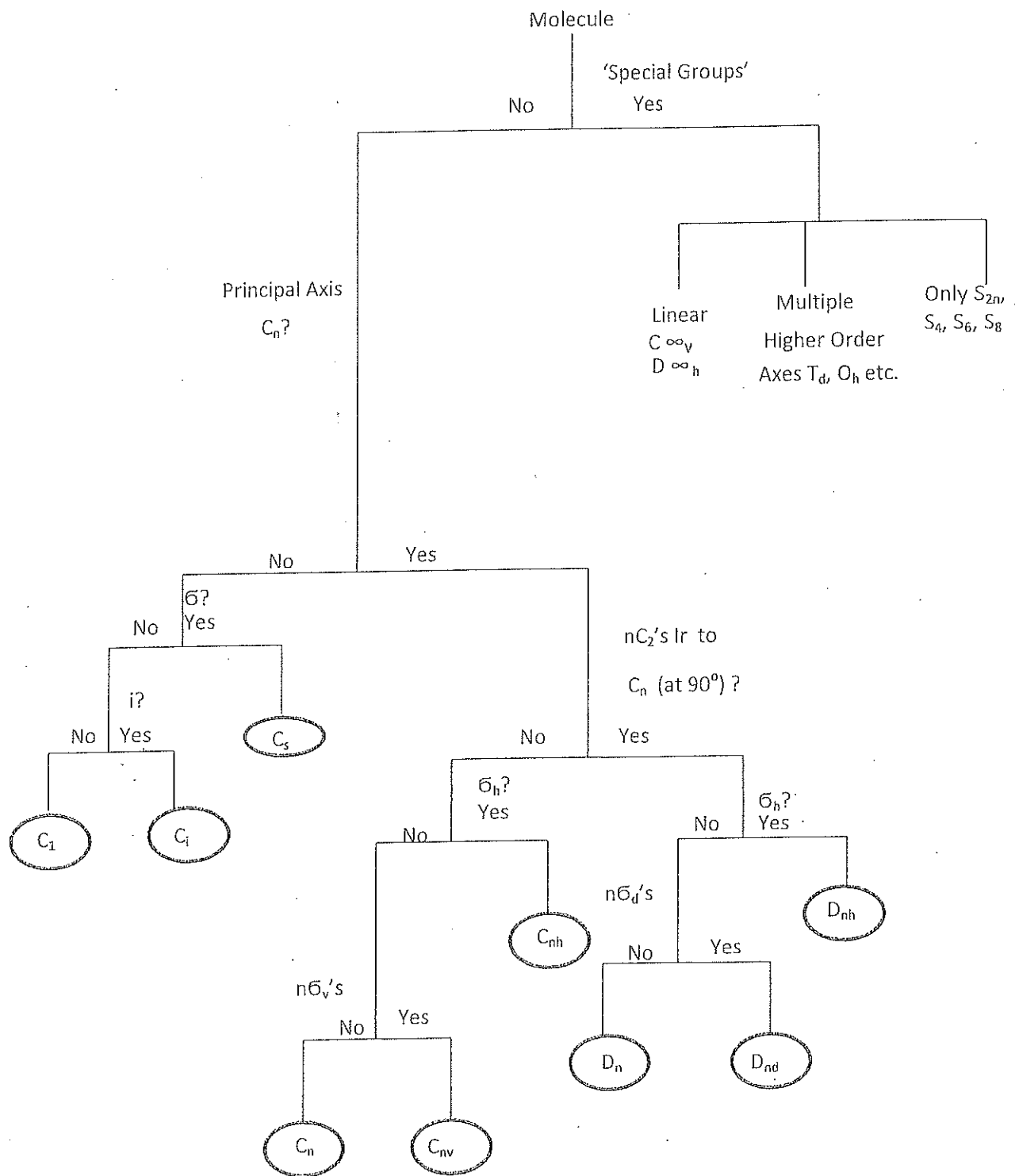
State the single symmetry operation of XeF_4 which has the same effect as:

- (1) S_4^2 (2) S_4^4 (3) C_4^2 (4) C_4^3 (5) σ^2 [5]
- (ii) For the following basis find the character representation. [5]



- (c) (i) Find the atomic orbitals on the central atom for bonding with the ligands in $[\text{FeCl}_4]^-$ (tetrahedral). [5]
- (ii) Hence construct a qualitative molecular orbital energy level diagram for σ -bonding. [5]

FLOW CHART FOR CLASSIFICATION OF POINT GROUPS.



Note: $C_{\infty v}$: Anti-symmetrical molecules e.g. HCN

$D_{\infty h}$: Symmetrical molecules e.g. CO_2

C_1 : No C_n or S_n , No σ and No i .

C_s : No C_n or S_n , but has σ .

C_i : No C_n or S_n , No σ but has i .

CONTRIBUTIONS BY VARIOUS SYMMETRY
OPERATIONS ON UNSHIFTED ATOM TO THE
CHARACTER

E	σ	i	C_n	S_n
3	1	-3	$2\cos\theta + 1$	$2\cos\theta - 1$
C_2	C_3	C_4	C_5	C_6
-1	0	1	1.618	2
S_3	S_4	S_5	S_6	S_8
-2	-1	-0.382	0	0.414

TRANSFORMATION OF SPECTROSCOPIC TERMS
INTO MULLIKEN SYMBOLS

Term	O_h	T_d
S	A_{1g}	A_1
P	T_{1g}	T_1
D	$E_g + T_{2g}$	$E + T_2$
F	$A_{2g} + T_{1g} + T_{2g}$	$A_2 + T_1 + T_2$
G	$A_{1g} + E_g + T_{1g} + T_{2g}$	$A_1 + E + T_1 + T_2$

PERIODIC TABLE OF ELEMENTS

GROUPS

PERIODS	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	
	IA	IIA	IIIB	IVB	VB	VIB	VIIA	VIII	VIII	X	IB	IIB	IIIA	IVA	VA	VIA	VIIA	VIIIA	
1	H 1																	He 2	
2	Li 3	Be 4																F 9	Ne 10
3	Na 11	Mg 12																Cl 17	Ar 18
TRANSITION ELEMENTS																			
4	K 19	Ca 20	Sc 21	Ti 22	V 23	Cr 24	Mn 25	Fe 26	Co 27	Ni 28	Cu 29	Zn 30	Ga 31	Ge 32	As 33	Se 34	Br 35	Kr 36	
5	Rb 37	Sr 38	Y 39	Zr 40	Nb 41	Mo 42	Tc 43	Ru 44	Rh 45	Pd 46	Ag 47	Cd 48	In 49	Sn 50	Sb 51	Te 52	I 53	Xe 54	
6	Cs 55	Ba 56	*La 57	Hf 72	Ta 73	W 74	Re 75	Os 76	Ir 77	Pt 78	Au 79	Hg 80	Tl 81	Pb 82	Bi 83	Po 84	At 85	Rn 86	
7	Rf 87	Ra 88	**Ac 89	Rf 104	Ha 105	Uuh 106	Uus 107	Uuo 108	Uue 109	Uun 110									

Atomic mass
Symbol
Atomic No.

140.12	140.91	144.24	(145)	150.36	151.96	157.25	158.93	162.50	164.93	167.26	168.93	173.04	174.97
Ce 58	Rf 59	Nd 60	Pm 61	Sm 62	Eu 63	Gd 64	Tb 65	Dy 66	Ho 67	Er 68	Tm 69	Yb 70	Lu 71
232.04	231.04	238.03	237.05	(244)	(243)	(247)	(251)	(252)	(257)	(258)	(259)	(260)	
Th 90	Pa 91	U 92	Np 93	Pu 94	Am 95	Cm 96	Bk 97	Cf 98	Es 99	Fm 100	Md 101	No 102	Lr 103

() indicates the mass number of the isotope with the longest half-life.

*Lanthanide Series

**Actinide Series

Character Tables for Chemically Important Symmetry Groups

1. The Nonaxial Groups

C_1	E
A	1

C_s	E	σ_h		
A'	1	1	x, y, R_z	x^2, y^2, z^2, xy
A''	1	-1	z, R_x, R_y	yz, xz

C_i	E	i		
A_g	1	1	R_x, R_y, R_z	$x^2, y^2, z^2, xy, xz, yz$
A_u	1	-1	x, y, z	

2. The C_n Groups

C_2	E	C_2		
A	1	1	z, R_z	x^2, y^2, z^2, xy
B	1	-1	x, y, R_x, R_y	yz, xz

C_3	E	C_3	C_3^2		$\epsilon = \exp(2\pi i/3)$
A	1	1	1	z, R_z	$x^2 + y^2, z^2$
E	$\begin{Bmatrix} 1 & \epsilon & \epsilon^* \\ 1 & \epsilon^* & \epsilon \end{Bmatrix}$			$(x, y)(R_x, R_y)$	$(x^2 - y^2, xy)(yz, xz)$

C_4	E	C_4	C_2	C_4^3		
A	1	1	1	1	z, R_z	$x^2 + y^2, z^2$
B	1	-1	1	-1		$x^2 - y^2, xy$
E	$\begin{Bmatrix} 1 & i & -1 & -i \\ 1 & -i & -1 & i \end{Bmatrix}$				$(x, y)(R_x, R_y)$	(yz, xz)

The C_n Groups (continued)

C_5	E	C_5	C_5^2	C_5^3	C_5^4		$\epsilon = \exp(2\pi i/5)$
A	1	1	1	1	1	z, R_z	$x^2 + y^2, z^2$
E_1	$\left\{ \begin{array}{l} 1 \quad \epsilon \quad \epsilon^2 \quad \epsilon^{2*} \quad \epsilon^* \\ 1 \quad \epsilon^* \quad \epsilon^{2*} \quad \epsilon^2 \quad \epsilon \end{array} \right\}$					$(x, y)(R_x, R_y)$	(yz, xz)
E_2	$\left\{ \begin{array}{l} 1 \quad \epsilon^2 \quad \epsilon^* \quad \epsilon \quad \epsilon^{2*} \\ 1 \quad \epsilon^{2*} \quad \epsilon \quad \epsilon^* \quad \epsilon^2 \end{array} \right\}$						$(x^2 - y^2, xy)$

C_6	E	C_6	C_3	C_2	C_3^2	C_6^5		$\epsilon = \exp(2\pi i/6)$
A	1	1	1	1	1	1	z, R_z	$x^2 + y^2, z^2$
B	1	-1	1	-1	1	-1		
E_1	$\left\{ \begin{array}{l} 1 \quad \epsilon \quad -\epsilon^* \quad -1 \quad -\epsilon \quad \epsilon^* \\ 1 \quad \epsilon^* \quad -\epsilon \quad -1 \quad -\epsilon^* \quad \epsilon \end{array} \right\}$						(x, y) (R_x, R_y)	(xz, yz)
E_2	$\left\{ \begin{array}{l} 1 \quad -\epsilon^* \quad -\epsilon \quad 1 \quad -\epsilon^* \quad -\epsilon \\ 1 \quad -\epsilon \quad -\epsilon^* \quad 1 \quad -\epsilon \quad -\epsilon^* \end{array} \right\}$							$(x^2 - y^2, xy)$

C_7	E	C_7	C_7^2	C_7^3	C_7^4	C_7^5	C_7^6		$\epsilon = \exp(2\pi i/7)$
A	1	1	1	1	1	1	1	z, R_z	$x^2 + y^2, z^2$
E_1	$\left\{ \begin{array}{l} 1 \quad \epsilon \quad \epsilon^2 \quad \epsilon^3 \quad \epsilon^{3*} \quad \epsilon^{2*} \quad \epsilon^* \\ 1 \quad \epsilon^* \quad \epsilon^{2*} \quad \epsilon^{3*} \quad \epsilon^3 \quad \epsilon^2 \quad \epsilon \end{array} \right\}$							(x, y) (R_x, R_y)	(xz, yz)
E_2	$\left\{ \begin{array}{l} 1 \quad \epsilon^2 \quad \epsilon^{3*} \quad \epsilon^* \quad \epsilon \quad \epsilon^3 \quad \epsilon^{2*} \\ 1 \quad \epsilon^{2*} \quad \epsilon^3 \quad \epsilon \quad \epsilon^* \quad \epsilon^{3*} \quad \epsilon^2 \end{array} \right\}$								$(x^2 - y^2, xy)$
E_3	$\left\{ \begin{array}{l} 1 \quad \epsilon^3 \quad \epsilon^* \quad \epsilon^2 \quad \epsilon^{2*} \quad \epsilon \quad \epsilon^{3*} \\ 1 \quad \epsilon^{3*} \quad \epsilon \quad \epsilon^{2*} \quad \epsilon^2 \quad \epsilon^* \quad \epsilon^3 \end{array} \right\}$								

C_8	E	C_8	C_4	C_2	C_4^3	C_8^3	C_8^5	C_8^7		$\epsilon = \exp(2\pi i/8)$
A	1	1	1	1	1	1	1	1	z, R_z	$x^2 + y^2, z^2$
B	1	-1	1	1	1	-1	-1	-1		
E_1	$\left\{ \begin{array}{l} 1 \quad \epsilon \quad i \quad -1 \quad -i \quad -\epsilon^* \quad -\epsilon \quad \epsilon^* \\ 1 \quad \epsilon^* \quad -i \quad -1 \quad i \quad -\epsilon \quad -\epsilon^* \quad \epsilon \end{array} \right\}$								(x, y) (R_x, R_y)	(xz, yz)
E_2	$\left\{ \begin{array}{l} 1 \quad i \quad -1 \quad 1 \quad -1 \quad -i \quad i \quad -i \\ 1 \quad -i \quad -1 \quad 1 \quad -1 \quad i \quad -i \quad i \end{array} \right\}$									$(x^2 - y^2, xy)$
E_3	$\left\{ \begin{array}{l} 1 \quad -\epsilon \quad i \quad -1 \quad -i \quad \epsilon^* \quad \epsilon \quad -\epsilon^* \\ 1 \quad -\epsilon^* \quad -i \quad -1 \quad i \quad \epsilon \quad \epsilon^* \quad -\epsilon \end{array} \right\}$									

3. The D_n Groups

D_2	E	$C_2(z)$	$C_2(y)$	$C_2(x)$			
A	1	1	1	1		x^2, y^2, z^2	
B_1	1	1	-1	-1	z, R_z	xy	
B_2	1	-1	1	-1	y, R_y	xz	
B_3	1	-1	-1	1	x, R_x	yz	
D_3	E	$2C_3$	$3C_2$				
$A_{1,2}$	1	1	1			$x^2 + y^2, z^2$	
A_2	1	1	-1		z, R_z		
E	2	-1	0		$(x, y)(R_x, R_y)$	$(x^2 - y^2, xy)(xz, yz)$	
D_4	E	$2C_4$	$C_2(=C_4^2)$	$2C_2'$	$2C_2''$		
A_1	1	1	1	1	1	$x^2 + y^2, z^2$	
A_2	1	1	1	-1	-1	z, R_z	
B_1	1	-1	1	1	-1	$x^2 - y^2$	
B_2	1	-1	1	-1	1	xy	
E	2	0	-2	0	0	$(x, y)(R_x, R_y)$ (xz, yz)	
D_5	E	$2C_5$	$2C_5^2$	$5C_2$			
A_1	1	1	1	1		$x^2 + y^2, z^2$	
A_2	1	1	1	-1		z, R_z	
E_1	2	$2 \cos 72^\circ$	$2 \cos 144^\circ$	0		(xz, yz)	
E_2	2	$2 \cos 144^\circ$	$2 \cos 72^\circ$	0		$(x^2 - y^2, xy)$	
D_6	E	$2C_6$	$2C_3$	C_2	$3C_2'$	$3C_2''$	
A_1	1	1	1	1	1	1	$x^2 + y^2, z^2$
A_2	1	1	1	1	-1	-1	z, R_z
B_1	1	-1	1	-1	1	-1	
B_2	1	-1	1	-1	-1	1	
E_1	2	1	-1	-2	0	0	(xz, yz)
E_2	2	-1	-1	2	0	0	$(x^2 - y^2, xy)$

4. The C_{nv} Groups

C_{2v}	E	C_2	$\sigma_v(xz)$	$\sigma'_v(yz)$		
A_1	1	1	1	1	z	x^2, y^2, z^2
A_2	1	1	-1	-1	R_z	xy
B_1	1	-1	1	-1	x, R_y	xz
B_2	1	-1	-1	1	y, R_x	yz

C_{3v}	E	$2C_3$	$3\sigma_v$			
A_1	1	1	1	z		$x^2 + y^2, z^2$
A_2	1	1	-1	R_z		
E	2	-1	0	$(x, y)(R_x, R_y)$		$(x^2 - y^2, xy)(xz, yz)$

C_{4v}	E	$2C_4$	C_2	$2\sigma_v$	$2\sigma_d$		
A_1	1	1	1	1	1	z	$x^2 + y^2, z^2$
A_2	1	1	1	-1	-1	R_z	
B_1	1	-1	1	1	-1		$x^2 - y^2$
B_2	1	-1	1	-1	1		xy
E	2	0	-2	0	0	$(x, y)(R_x, R_y)$	(xz, yz)

C_{5v}	E	$2C_5$	$2C_5^2$	$5\sigma_v$		
A_1	1	1	1	1	z	$x^2 + y^2, z^2$
A_2	1	1	1	-1	R_z	
E_1	2	$2 \cos 72^\circ$	$2 \cos 144^\circ$	0	$(x, y)(R_x, R_y)$	(xz, yz)
E_2	2	$2 \cos 144^\circ$	$2 \cos 72^\circ$	0		$(x^2 - y^2, xy)$

C_{6v}	E	$2C_6$	$2C_3$	C_2	$3\sigma_v$	$3\sigma_d$		
A_1	1	1	1	1	1	1	z	$x^2 + y^2, z^2$
A_2	1	1	1	1	-1	-1	R_z	
B_1	1	-1	1	-1	1	-1		
B_2	1	-1	1	-1	-1	1		
E_1	2	1	-1	-2	0	0	$(x, y)(R_x, R_y)$	(xz, yz)
E_2	2	-1	-1	2	0	0		$(x^2 - y^2, xy)$

6. The D_{nh} Groups

D_{2h}	E	$C_2(z)$	$C_2(y)$	$C_2(x)$	i	$\sigma(xy)$	$\sigma(xz)$	$\sigma(yz)$		
A_g	1	1	1	1	1	1	1	1		x^2, y^2, z^2
B_{1g}	1	1	-1	-1	1	1	-1	-1	R_x	xy
B_{2g}	1	-1	1	-1	1	-1	1	-1	R_y	xz
B_{3g}	1	-1	-1	1	1	-1	-1	1	R_z	yz
A_u	1	1	1	1	-1	-1	-1	-1		
B_{1u}	1	1	-1	-1	-1	-1	1	1	z	
B_{2u}	1	-1	1	-1	-1	1	-1	1	y	
B_{3u}	1	-1	-1	1	-1	1	1	-1	x	

D_{3h}	E	$2C_3$	$3C_2$	σ_h	$2S_6$	$3\sigma_v$		
A_1'	1	1	1	1	1	1		$x^2 + y^2, z^2$
A_2'	1	1	-1	1	1	-1	R_x	
E'	2	-1	0	2	-1	0	(x, y)	$(x^2 - y^2, xy)$
A_1''	1	1	1	-1	-1	-1		
A_2''	1	1	-1	-1	-1	1	z	
E''	2	-1	0	-2	1	0	(R_x, R_y)	(xz, yz)

D_{4h}	E	$2C_4$	C_2	$2C_2'$	$2C_2''$	i	$2S_4$	σ_h	$2\sigma_v$	$2\sigma_d$		
A_{1g}	1	1	1	1	1	1	1	1	1	1		$x^2 + y^2, z^2$
A_{2g}	1	1	1	-1	-1	1	1	1	-1	-1	R_x	
B_{1g}	1	-1	1	1	-1	1	-1	1	1	-1		$x^2 - y^2$
B_{2g}	1	-1	1	-1	1	1	-1	1	-1	1		xy
E_g	2	0	-2	0	0	2	0	-2	0	0	(R_x, R_y)	(xz, yz)
A_{1u}	1	1	1	1	1	-1	-1	-1	-1	-1		
A_{2u}	1	1	1	-1	-1	-1	-1	-1	1	1	z	
B_{1u}	1	-1	1	1	-1	-1	1	-1	-1	1		
B_{2u}	1	-1	1	-1	1	-1	1	-1	1	-1		
E_u	2	0	-2	0	0	-2	0	2	0	0	(x, y)	

D_{5h}	E	$2C_5$	$2C_5^2$	$5C_2$	σ_h	$2S_5$	$2S_5^3$	$5\sigma_v$		
A_1'	1	1	1	1	1	1	1	1		$x^2 + y^2, z^2$
A_2'	1	1	1	-1	1	1	1	-1	R_x	
E_1'	2	$2 \cos 72^\circ$	$2 \cos 144^\circ$	0	2	$2 \cos 72^\circ$	$2 \cos 144^\circ$	0	(x, y)	
E_2'	2	$2 \cos 144^\circ$	$2 \cos 72^\circ$	0	2	$2 \cos 144^\circ$	$2 \cos 72^\circ$	0		$(x^2 - y^2, xy)$
A_1''	1	1	1	1	-1	-1	-1	-1		
A_2''	1	1	1	-1	-1	-1	-1	1	z	
E_1''	2	$2 \cos 72^\circ$	$2 \cos 144^\circ$	0	-2	$-2 \cos 72^\circ$	$-2 \cos 144^\circ$	0	(R_x, R_y)	(xz, yz)
E_2''	2	$2 \cos 144^\circ$	$2 \cos 72^\circ$	0	-2	$-2 \cos 144^\circ$	$-2 \cos 72^\circ$	0		

D_{6h}	E	$2C_6$	$2C_3$	C_2	$3C_2'$	$3C_2''$	i	$2S_6$	$2S_6^5$	σ_h	$3\sigma_d$	$3\sigma_v$		
A_{1g}	1	1	1	1	1	1	1	1	1	1	1	1		$x^2 + y^2, z^2$
A_{2g}	1	1	1	1	-1	-1	1	1	1	1	-1	-1	R_x	
B_{1g}	1	-1	1	-1	1	-1	1	-1	1	-1	1	-1		
B_{2g}	1	-1	1	-1	-1	1	1	-1	1	-1	-1	1		
E_{1g}	2	1	-1	-2	0	0	2	1	-1	-2	0	0	(R_x, R_y)	(xz, yz)
E_{2g}	2	-1	-1	2	0	0	2	-1	-1	2	0	0		$(x^2 - y^2, xy)$
A_{1u}	1	1	1	1	1	1	-1	-1	-1	-1	-1	-1		
A_{2u}	1	1	1	1	-1	-1	-1	-1	-1	-1	1	1	z	
B_{1u}	1	-1	1	-1	1	-1	-1	1	-1	1	-1	1		
B_{2u}	1	-1	1	-1	-1	1	-1	1	-1	1	1	-1		
E_{1u}	2	1	-1	-2	0	0	-2	-1	1	2	0	0	(x, y)	
E_{2u}	2	-1	-1	2	0	0	-2	1	1	-2	0	0		

7. The D_{nd} Groups

D_{2d}	E	$2S_4$	C_2	$2C_2'$	$2\sigma_d$		
A_1	1	1	1	1	1	R_z	$x^2 + y^2, z^2$
A_2	1	1	1	-1	-1		$x^2 - y^2$
B_1	1	-1	1	1	-1		xy
B_2	1	-1	1	-1	1		(xz, yz)
E	2	0	-2	0	0		$(x, y);$ (R_x, R_y)

D_{3d}	E	$2C_3$	$3C_2$	i	$2S_6$	$3\sigma_d$			
A_{1g}	1	1	1	1	1	1	R_z	$x^2 + y^2, z^2$	
A_{2g}	1	1	-1	1	1	-1		(R_x, R_y)	$(x^2 - y^2, xy)$
E_g	2	-1	0	2	-1	0		(xz, yz)	
A_{1u}	1	1	1	-1	-1	-1	z		
A_{2u}	1	1	-1	-1	-1	1		(x, y)	
E_u	2	-1	0	-2	1	0			

D_{4d}	E	$2S_4$	$2C_4$	$2S_4^3$	C_2	$4C_2'$	$4\sigma_d$		
A_1	1	1	1	1	1	1	1	R_z	$x^2 + y^2, z^2$
A_2	1	1	1	1	1	-1	-1		
B_1	1	-1	1	-1	1	1	-1		z
B_2	1	-1	1	-1	1	-1	1		(x, y)
E_1	2	$\sqrt{2}$	0	$-\sqrt{2}$	-2	0	0	(R_x, R_y)	$(x^2 - y^2, xy)$
E_2	2	0	-2	0	2	0	0		(xz, yz)
E_3	2	$-\sqrt{2}$	0	$\sqrt{2}$	-2	0	0		

D_{5d}	E	$2C_5$	$2C_5^2$	$5C_2$	i	$2S_{10}$	$2S_{10}^3$	$5\sigma_d$			
A_{1g}	1	1	1	1	1	1	1	1	R_z	$x^2 + y^2, z^2$	
A_{2g}	1	1	1	-1	1	1	1	-1		(R_x, R_y)	(xz, yz)
E_{1g}	2	$2 \cos 72^\circ$	$2 \cos 144^\circ$	0	2	$2 \cos 72^\circ$	$2 \cos 144^\circ$	0			$(x^2 - y^2, xy)$
E_{2g}	2	$2 \cos 144^\circ$	$2 \cos 72^\circ$	0	2	$2 \cos 144^\circ$	$2 \cos 72^\circ$	0			
A_{1u}	1	1	1	1	-1	-1	-1	-1		z	
A_{2u}	1	1	1	-1	-1	-1	-1	1	(x, y)		
E_{1u}	2	$2 \cos 72^\circ$	$2 \cos 144^\circ$	0	-2	$-2 \cos 72^\circ$	$-2 \cos 144^\circ$	0			
E_{2u}	2	$2 \cos 144^\circ$	$2 \cos 72^\circ$	0	-2	$-2 \cos 144^\circ$	$-2 \cos 72^\circ$	0			

D_{6d}	E	$2S_{12}$	$2C_6$	$2S_4$	$2C_3$	$2S_{12}^5$	C_2	$6C_2'$	$6\sigma_d$		
A_1	1	1	1	1	1	1	1	1	1	R_z	$x^2 + y^2, z^2$
A_2	1	1	1	1	1	1	1	-1	-1		
B_1	1	-1	1	-1	1	-1	1	1	-1		z
B_2	1	-1	1	-1	1	-1	1	-1	1		(x, y)
E_1	2	$\sqrt{3}$	1	0	-1	$-\sqrt{3}$	-2	0	0	(R_x, R_y)	$(x^2 - y^2, xy)$
E_2	2	1	-1	-2	-1	1	2	0	0		
E_3	2	0	-2	0	2	0	-2	0	0		
E_4	2	-1	-1	2	-1	-1	2	0	0		
E_5	2	$-\sqrt{3}$	1	0	-1	$\sqrt{3}$	-2	0	0		

8. The S_n Groups

S_4	E	S_4	C_2	S_4^3		
A	1	1	1	1	R_z	$x^2 + y^2, z^2$
B	1	-1	1	-1	z	$x^2 - y^2, xy$
E	$\begin{Bmatrix} 1 & i & -1 & -i \\ 1 & -i & -1 & i \end{Bmatrix}$				$(x, y); (R_x, R_y)$	(xz, yz)

S_6	E	C_3	C_3^2	i	S_6^5	S_6		$\epsilon = \exp(2\pi i/3)$
A_z	1	1	1	1	1	1	R_z	$x^2 + y^2, z^2$
E_ϵ	$\begin{Bmatrix} 1 & \epsilon & \epsilon^* & 1 & \epsilon & \epsilon^* \\ 1 & \epsilon^* & \epsilon & 1 & \epsilon^* & \epsilon \end{Bmatrix}$						(R_x, R_y)	$(x^2 - y^2, xy);$ (xz, yz)
A_u	1	1	1	-1	-1	-1	z	
E_u	$\begin{Bmatrix} 1 & \epsilon & \epsilon^* & -1 & -\epsilon & -\epsilon^* \\ 1 & \epsilon^* & \epsilon & -1 & -\epsilon^* & -\epsilon \end{Bmatrix}$						(x, y)	

S_8	E	S_8	C_4	S_8^3	C_2	S_8^5	C_4^3	S_8^7		$\epsilon = \exp(2\pi i/8)$
A	1	1	1	1	1	1	1	1	R_z	$x^2 + y^2, z^2$
B	1	-1	1	-1	1	-1	1	-1	z	
E_1	$\begin{Bmatrix} 1 & \epsilon & i & -\epsilon^* & -1 & -\epsilon & -i & \epsilon^* \\ 1 & \epsilon^* & -i & -\epsilon & -1 & -\epsilon^* & i & \epsilon \end{Bmatrix}$								$(x, y);$ (R_x, R_y)	
E_2	$\begin{Bmatrix} 1 & i & -1 & -i & 1 & i & -1 & -i \\ 1 & -i & -1 & i & 1 & -i & -1 & i \end{Bmatrix}$									$(x^2 - y^2, xy)$
E_3	$\begin{Bmatrix} 1 & -\epsilon^* & -i & \epsilon & -1 & \epsilon^* & i & -\epsilon \\ 1 & -\epsilon & i & \epsilon^* & -1 & \epsilon & -i & -\epsilon^* \end{Bmatrix}$									(xz, yz)

11. The Icosahedral Group

I_A	E	$12C_5$	$12C_5^2$	$20C_3$	$15C_2$	i	$12S_{10}$	$12S_{10}^2$	$20S_4$	15σ	
A_1	1	1	1	1	1	1	1	1	1	1	$x^2 + y^2 + z^2$
T_{1F}	3	$\frac{1}{2}(1 + \sqrt{5})$	$\frac{1}{2}(1 - \sqrt{5})$	0	-1	3	$\frac{1}{2}(1 - \sqrt{5})$	$\frac{1}{2}(1 + \sqrt{5})$	0	-1	$(2x^2 - x^2 - y^2, x^2 - y^2, xy, yz, zx)$
T_{2F}	3	$\frac{1}{2}(1 - \sqrt{5})$	$\frac{1}{2}(1 + \sqrt{5})$	0	-1	3	$\frac{1}{2}(1 + \sqrt{5})$	$\frac{1}{2}(1 - \sqrt{5})$	0	-1	
G_F	4	-1	-1	1	0	4	-1	1	1	0	
H_6	5	0	0	-1	1	5	0	-1	-1	1	
A_u	1	1	1	1	1	1	1	1	1	1	
T_{1u}	3	$\frac{1}{2}(1 + \sqrt{5})$	$\frac{1}{2}(1 - \sqrt{5})$	0	-1	-3	$-\frac{1}{2}(1 - \sqrt{5})$	$-\frac{1}{2}(1 + \sqrt{5})$	0	1	(x, y, z)
T_{2u}	3	$\frac{1}{2}(1 - \sqrt{5})$	$\frac{1}{2}(1 + \sqrt{5})$	0	-1	-3	$-\frac{1}{2}(1 + \sqrt{5})$	$-\frac{1}{2}(1 - \sqrt{5})$	0	1	
G_u	4	-1	-1	1	0	-4	1	1	-1	0	
H_u	5	0	0	-1	1	-5	0	-1	1	-1	